

Total No. of Questions :10]

SEAT No. :

P4233

[4860]-617-A

[Total No. of Pages :4

M.E. (Civil) (Water Resources and Environmental Engineering)

INDUSTRIAL WASTE MANAGEMENT

(2012 Course) (Semester - II) (Elective -III) (501611)

Time : 3 Hours]

[Max. Marks :100

Instructions to the candidates:

- 1) Answers to the two sections should be written in separate answer books.
- 2) Attempt any three questions from each sections.
- 3) Neat diagrams must be drawn wherever necessary.
- 4) Figures to the right side indicate full marks.
- 5) Use of calculator is allowed.
- 6) Assume suitable data, if necessary.
- 7) Use data sheet.

**SECTION-I**

**Q1)** a) Explain physical, chemical & biological characteristics of industrial waste water. [8]

b) The cumulative flow of waste water (W.W.) reaching to CETP in a day varies as shown in table No.1 Determine the capacity of equalization tank for given flow variations. Plot graph of time Vs Cumulative flow.[8]

Table No.1 Data for finding capacity of equalization tank

Time Hr.	0	2	4	6	8	10	12	14	16	18	20	22	24
Cumulative flow m <sup>3</sup>	0	25	50	75	100	110	130	140	150	165	170	198	220

**Q2)** a) Explain grab and composite sampling technique. [8]

b) Explain techniques for reduction of waste. [8]

**P.T.O.**

**Q3) a)** Plot the BOD curve and calculate  $K_t$  &  $L_t$  for the following data. [8]

Time days	2	4	6	8	10
BOD mg/L	10	20	23	25	30

b) Explain the concept of self purification of stream. [8]

**Q4) a)** Explain different methods for removal of TDS. [8]

b) Design an aerobic sludge digester for Industrial waste water of 10 MLD, quantity of sludge produced = 2000 kg/d, Sp. Gravity = 1.003, Solid concentration = 3.5%, HRT = 15 days, Volatile solids = 80% Find: [8]

i) Volume of sludge digested per day.

ii) Volume of sludge digester.

iii) Oxygen required take oxygen requirement as 2 kg/kg-d of volatile solid cell oxidized and 40% cell destroyed.

iv) Volume of air supplied  $1.20 \times 0.21 \text{ kg/m}^3$ .

**Q5) a)** The BOD of industrial W.W. for 1 day at 30°C has been found to be 110 mg/L. What will be the 5 day BOD at 20°C. Take  $K_D(20^\circ\text{C}) = 0.1$ . [9]

b) Explain the concept of anaerobic sludge digestion with three phases such as hydrolysis, acidogenesis, methanogenesis and also explain conventional sludge digester with diagram. [9]

## **SECTION-II**

**Q6) a)** The treated W.W. with flow of 2000 L/min is to be treated with PAC to reduced concentration of total organic nitrogen from 5 mg/L to 1 mg/L. The removal follows Freundlich adsorption isotherm in which capacity factor and intensity parameter are  $160 \text{ mg/gm (L/mg)}^{1/n}$  & 0.5 respectively. Determine PAC dose, PAC requirement and annual cost of PAC. Assume PAC cost to be Rs. 50/- per kg. [8]

b) Explain operation and maintenance problem of CETP. [8]

- Q7) a)** Draw a flow diagram for treating W.W. of dairy industry & show all treatment unit. [8]
- b) Draw a flow diagram for treating W.W. of textile industry and show all treatment unit. [8]
- Q8) a)** Explain the electro dialysis process and determine power required for electro dialysis process for the following data: [8]
- i)  $Q = 4000 \text{ m}^3/\text{d}$ ,
  - ii)  $\text{TDS} = 2000 \text{ mg/L}$ ,
  - iii) No. of Cells = 300
  - iv) Cation and anions concentration =  $0.011 \text{ Eq/L}$ ,
  - v) Salt removal Efficiency = 50%, Current efficiency = 90%.,  $R = 50 \text{ ohm}$ . Use data sheet if required.
- b) Explain importance of treatability index in waste water treatment. [8]
- Q9) a)** Explain the characteristics of sugar industrial waste with value and state disposal method of industrial treated waste with flow diagram. [8]
- b) Write short note on water pollution act 1978. [8]
- Q10)a)** Design CETP for following data [9]
- i)  $Q = 150 \text{ m}^3/\text{d}$ ,
  - ii)  $\text{pH} = 11 - 12$ ,
  - iii)  $\text{BOD at } 27^\circ\text{C} = 70 \text{ mg/L}$
  - iv)  $\text{TDS} = 3000 \text{ mg/L}$
  - v)  $\text{TS} = 6000 \text{ mg/L}$
  - vi) Iron concentration =  $700 \text{ mg/L}$
  - vii)  $\text{COD} = 5000 \text{ mg/L at } 27^\circ\text{C}$

Find:

- 1) Volume of sedimentation tank
  - 2) Acid required for neutralization if 3 lit acid required for 1 m<sup>3</sup>/d of E/F pH.
  - 3) Quantity of lime required for removal of Iron take lime consumption as 1mg/L of Iron required 0.5 mg/L of lime.
  - 4) Select suitable method for removal of COD and BOD.
  - 5) F/M ratio
- b) Write short note on hazardous waste management. [9]

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