

Total No. of Questions : 8]

SEAT No. :

P4119

[Total No. of Pages : 3

[4760] - 1047

M.E. (Civil) (Water Resources and Environment Engineering)
ENVIRONMENTAL CHEMISTRY & MICROBIOLOGY
(2013 Pattern) (Semester - I)

Time : 3 Hours]

[Max. Marks : 50

Instructions to the candidates:

- 1) Attempt any five questions.*
- 2) Neat diagrams must be drawn wherever necessary.*
- 3) Figures to the right side indicate full marks.*
- 4) Use of calculator is allowed.*
- 5) Assume suitable data if necessary.*
- 6) Use data sheet.*

- Q1)** a) State the stability of atmosphere for the following data Temp. of air at ground = 30°C. Temp. of air at 200m height = 20°C. If a balloon is release from a ground after 1 hour reach to height of 100m at that time temperature of 22°C. **[4]**
- b) Explain radiation and types of radiation. **[3]**
- c) A high volume sampler is run for 30 min @ 20m³/min air. Initial weight of filter paper is 5 gm after 30 min weight of filter paper is 10 gm what is concentration of dust particle in microgram/m³. **[3]**
- Q2)** a) Design multi cyclone chamber for flue gas of 10m³/sec. Assume all the necessary data. **[4]**
- b) A fabric filter must process 3m³/s of flue gas. Design the bag house filter with air to cloth ratio of 4m³/min/m² Determine no. of bags and physical arrangement. Take dia. of each base as 200mm. **[3]**
- c) Explain working of adsorption process in details. **[3]**

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Q3) a) A gas has density of 1.89 kg/m^3 at a pressure of 1 bar with temperature of 30°C . A mass of 0.9 kg of the gas requires a heat transfer of 300 kJ to raise the temperature. Determine. [3]

i) Characteristic gas constant.

ii) C_p of the Gas

iii) C_v of the Gas

Change in internal energy of gas

b) Explain steam curve. [3]

c) Explain vapor absorption system with diagram. [4]

Q4) a) Determine reaction order of reactant removal for the data given in table : [4]

concentration	230	180	110	60	40	22
Time (minute)	0	9	20	30	42	33

b) Explain working & principle of ion exchange process. [3]

c) A cold fluid flows through a heat exchanger at a rate of $20 \text{ m}^3/\text{hr}$. The temperature at entry of fluid and exit is 38°C and 62°C . The other fluid enters at a rate of $30 \text{ m}^3/\text{hr}$ and temperature is 112°C . Find the area required for heat exchanger in parallel and counter current flow conditions. [3]

Q5) a) Write short note on toxicity test. [3]

b) Explain the concept of anaerobic sludge digestion with three phases such as hydrolysis, acidogenesis, methanogenesis and also explain conventional sludge digester with diagram. [4]

c) A one cubic meter of air was found to contain $100 \text{ microgram/m}^3$ of CO_2 . The temperature & pressure are 20°C and 103.12 kPa when the air sample was taken. What was the concentration of CO_2 in PPM. [3]

Q6) Explain working mechanism of : [10]

a) pH meter

b) Flame photometer

c) Atomic Absorption Spectrophotometer

- Q7)** a) What are different type of reactor used in water treatment process. [6]
- b) Design ASP for Industrial waste water of 10,000 m³/d, BOD of I/F = 200 mg/L, E/F BOD = 10 mg/L, $Y = 0.5$, $K_d = 0.05$ / day, MLSS = 4000 mg/L, Return sludge concentration = 10,000 mg/L, MCRT = 10 days [4]
- Determine 1. Volume of reactor
- i) F/M ration.
 - ii) OLR
 - iii) Oxygen requirement
 - iv) Recycle ratio
- Q8)** Explain uses of algae and problems caused by algae in water along with their controlling measures. [10]

