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F.E. (Common to All Branches) (Second Semester)

EXAMINATION, 2016

APPLIED SCIENCE—II

(Chemistry)

(2008 PATTERN)

Time : Two Hours

Maximum Marks : 50

N.B. :— (i) Answer *three* questions from Section I and *three* questions from Section II.

(ii) Neat diagrams must be drawn wherever necessary.

(iii) Figures to the right indicate full marks.

(iv) Use of logarithmic tables, side rule, Mollier charts, electronic pocket calculator and steam tables is allowed.

(v) Assume suitable data, if necessary.

1. (a) Give the principle of Bomb calorimeter. How gross calorific value of solid fuel is determined by Bomb calorimeter ? [7]
- (b) Give preparation reaction of biodiesel. State merits and demerits of biodiesel. [6]
- (c) 0.25 gm of a coal sample on burning in combustion apparatus in the current of pure oxygen was found to increase weight of U-tube with anhydrous CaCl_2 by 0.075 gm and of KOH U-tube by 0.52 gm. Calculate percentage of C and H in coal sample. [4]

P.T.O.

Or

- 2.** (a) Explain in brief the process of distillation of crude petroleum. Give composition, boiling range and uses of any *three* fractions obtained. [7]
- (b) What is rocket propellant ? Explain different types of propellant used in rocket. [6]
- (c) Calculate the weight of air required for complete combustion of 100 kg of coal if it contains C = 82%, H₂ = 6%, O₂ = 2%, S = 4%, remaining ash. [4]
- 3.** (a) What is principle of cathodic protection ? Discuss the various types of cathodic protection. [7]
- (b) Explain different factors affecting rate of corrosion. [6]
- (c) Give the names of oxide films in Mg, Cr, Ag and Mo metals. [4]

Or

- 4.** (a) Define corrosion. Explain the mechanism of wet corrosion by H₂ evolution and O₂ absorption. [7]
- (b) Explain galvanising and tinning methods for applying metallic coatings. [6]
- (c) Write a note on electroplating of metal. [4]
- 5.** (a) What is priming and foaming ? What are the disadvantages of priming and foaming ? How can they be prevented ? [6]

- (b) How chloride quantity in water is determined by Mohr's method ? [6]
- (c) 50 ml of an alkaline water sample requires 9.2 ml of N/50 HCl upto phenolphthalein end point and total 13.1 ml of the acid for complete neutralization. Find the types and amount of alkalinities in water. [4]

Or

6. (a) Draw and explain phase diagram for water system. [6]
- (b) Explain zeolite process for water softening. [6]
- (c) Give the differences in sludge and scale in boiler. [4]