Marking scheme: Engineering Chemistry

U113-1011 (ESE)

A.Y. 2018-19 SEM-I End Semester examination Set 3

Question	Sub question number	Marking Scheme
Q.1	a)	Reactions - 1 mark, factors - 1mark, 2 advantages - 1 mark, 2 disadvantages - 1 mark
	b)	principle – 1 mark, Quantity of water = 48077 lit – 3 marks
Q.2	a)	4 applications- 4 marks (1 mark each)
	b)	i) 239 nm ii) 249 nm- 2 marks each
Q.3	a)	Principle of distillation – 1 mark, figure – 2 marks, three fractions
	b)	Figure – 2 marks, construction – 1 mark, working – 1 mark, 2formulae – 2 marks
Q.4	a)	Definition-1 mark, 5 factors- 5 marks
	b)	functions of any four ingredients- 4 marks (1 mark each)
Q.5	a)	PHBV- 3 marks (structure-1 mark, properties-1mark, applications-1mark) Kevlar- 3 marks (structure-1 mark, properties-1mark, applications-1mark)
	b)	four points- 4 marks
Q.6	a)	Cell reactions – 1 mark, procedure – 1 mark, calculations – 3 marks, titration curve – 1 mark
	b)	Discharge reactions – 2 marks, application – 1 mark, advantages – 1 mark
Q.7	a)	construction with figure – 2 marks, cell representation – 1 mark, electrode reactions – 2 marks, two uses- 1 mark
	b)	Construction with figure – 2 marks, Working reactions- 1 marks, 2advantages – 1 mark
Q.8	a)	hydrogen evolution mechanism and oxygen absorption mechanism – 3 marks each(condition of electrolyte, reactions at anode and cathode-1 mark, figure – 1 mark, explanation with example- 1 mark
	b)	Any four factors with explanation – 4 marks
Q.9	a)	principle,figure,process,2advantages,2disadvantages and2applications of anodic protection - 1 mark each
	b)	Oxidation corrosion of Mg,Cr,Mo and Al -4Marks (1 marks each,Reaction-1/2 mark, type of oxide film-1/2mark)
Q.10	a)	(iv) All of the above
	b)	iii) 1000-1100°C
	c)	(i) 0.91 M HCI
	d)	(i) Cathodic
		(ii) Buried cables
	e)	(III) Buried capies