Total No. of Questions - [3]

G.R. No.

EC

Total No. of Printed Pages: 2

MARCH FEBRUARY 2020 / INSEM (T1) F. Y. B.TECH. (COMMON) (SEMESTER - II)

F. Y. B.TECH. (COMMON) (SEMESTER - COURSE NAME: Engineering Chemistry

COURSE CODE: ES10184B (PATTERN 2018)

Time: [1 Hour]

[Max. Marks: 20]

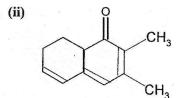
- (*) Instructions to candidates:
- 1) All questions are compulsory.
- 2) Figures to the right indicate full marks.
- 3) Use of scientific calculator is allowed.
- 4) Use suitable data where ever required.
- Q.1) Attempt any two
 - a) Compare zeolite and ion exchange treatment

[4]

- b) 50 ml of an alkaline water sample requires 3.7 ml of N/50 HCl upto phenolphthalein end point and further 4.8 ml for complete neutralization. Find the type and amount of alkalinity
- c) Explain any 4 methods of removal of micro-organisms in purification of water [4] for domestic use
- Q.2) Attempt any two
 - a) Calculate λ_{max} for the following compounds (Explain calculations):

[2+2]

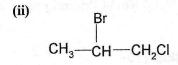
CH₃



- b) i) How can you distinguish between Ethanol and Acetone by IR [2+2] spectroscopy?
 - ii) Explain any 2 applications of IR spectroscopy
- c) Draw the low and high resolution spectra of the following

[2+2]

(i) CH₃—CH₂—CH₂—OH



Q.3) Attempt any one
a) What is Pilling Bedworth ratio? Discuss oxidation corrosion in case of Mg, [4] Cr, Mo with reactions and type of oxide film formed.
b) Explain electroplating with figure, process, reactions, 2 applications [4]

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