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PAPER CODE	P.122-223 QSE/PSE
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May 2022 / INSEM+ENDSEM
F. Y. M. TECH. (WREE-Civil) (SEMESTER - II)
COURSE NAME: Industrial Waste Water Treatment
COURSE CODE: CVPA12203
(PATTERN 2020)

Time: [3 Hours]

[Max. Marks: 60]

(*) Instructions to candidates:

- 1) All Questions are compulsory
- 2) Figures to the right indicate full marks.
- 3) Use of scientific calculator is allowed
- 4) Use suitable data where ever required

Q.1) a) What are physical and chemical characteristics of textile, dairy industries. [5]

b) Elaborate unit operation carried out in industrial waste water. [5]

Q.2)a) Estimate the volume of equalization tank required cum. in ETP for the following data. [5]

Q m ³ /min	12	18	22	29	35	39	42	52	60
Time (hr)	1	2	3	4	5	6	7	8	9

b) Discuss various methods used for removal of Chromium and Nickel from waste water. [5]

Q.3)a) Aerobic biological process is carried out in treatment of industrial waste water for the following parameters

I/F BOD = 800 mg/l, E/F BOD = 50 mg/L, $X = 5000 \text{ mg/l}$, $X_r = 10000 \text{ mg/l}$, HRT = 4.5 hr, MCRT = 10d, $K = 0.1/\text{day}$, Q(discharge) 20MLD, $Y = 0.5$, $K_d = 0.06/\text{day}$. Find Volume of tank, Y observed, Sludge production, BOD removal efficiency, F/M ratio, Volumetric loading rate. [5]

b) Explain Fluidized bed reactor working mechanism in anaerobic treatment system. [5]

Q.4)a) What are different method for removal of color from industrial waste water. [5]

b) Determine power required for electro dialysis process for the following data

1. $Q = 4000 \text{ m}^3/\text{d}$, 2. $\text{TDS} = 2000 \text{ mg/L}$, 3. No of Cells = 300
4. Cation and anions concentration = 0.011 Eq/L , 5. Salt removal Efficiency = 50%, Current efficiency = 90% ., $R = 50 \text{ ohm}$ [5]

Q.5) Design CETP for following data [10]

1. $Q = 50 \text{ m}^3/\text{d}$, 2. $\text{pH} = 11-12$, 3. $\text{BOD at } 27^\circ\text{C} = 70 \text{ mg/L}$
4. $\text{TDS} = 3000 \text{ mg/L}$ 5. $\text{TS} = 6000 \text{ mg/L}$ 6. Iron concentration = 700 mg/L
7. $\text{COD} = 5000 \text{ mg/L at } 27^\circ\text{C}$, 8. Color is present in waste water.

Find 1. Volume of sedimentation tank

2. Acid required for neutralization if 3 lit acid required for 1 m^3/d of E/F pH
3. Quantity of lime required for removal of Iron take lime consumption as 1 mg/L of Iron required 0.5 mg/L of lime
4. Select suitable method for removal of COD and BOD
5. F/M ratio.

Q.6) Draw flow diagram for manufacturing process and treatment process for textile, paper and pulp industries. Also write chemical composition in tabular form. [10]

*****End*****